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# $[\eta^2\text{-}8\text{-(}tert\text{-Butylimino)-}1\text{-naphthyl]bis}(\eta^5\text{-cyclopentadienyl)titanium(IV)$

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## Key indicators

Single-crystal X-ray study

$T = 298\text{ K}$

Mean  $\sigma(\text{C}-\text{C}) = 0.006\text{ \AA}$

$R$  factor = 0.056

$wR$  factor = 0.131

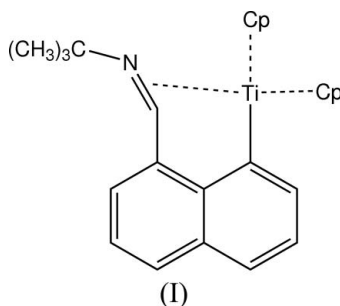
Data-to-parameter ratio = 18.4

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound,  $[\text{Ti}(\text{C}_5\text{H}_5)_2(\text{C}_{15}\text{H}_{15}\text{N})]$ , displays an iminoacyl fragment in an N-outside conformation, indicating that the  $\text{C}=\text{N}$  vector points away from the  $\sigma$ -bonded C atom. Considering only the geometrical centres of all  $\eta$ -coordinated groups, the  $\text{Ti}^{\text{IV}}$  ion adopts a severely distorted tetrahedral configuration. The crystal packing displays a short  $\text{C}-\text{H}\cdots\pi$  contact linking the molecules into an infinite chain running along the  $b$ -axis direction.

## Comment

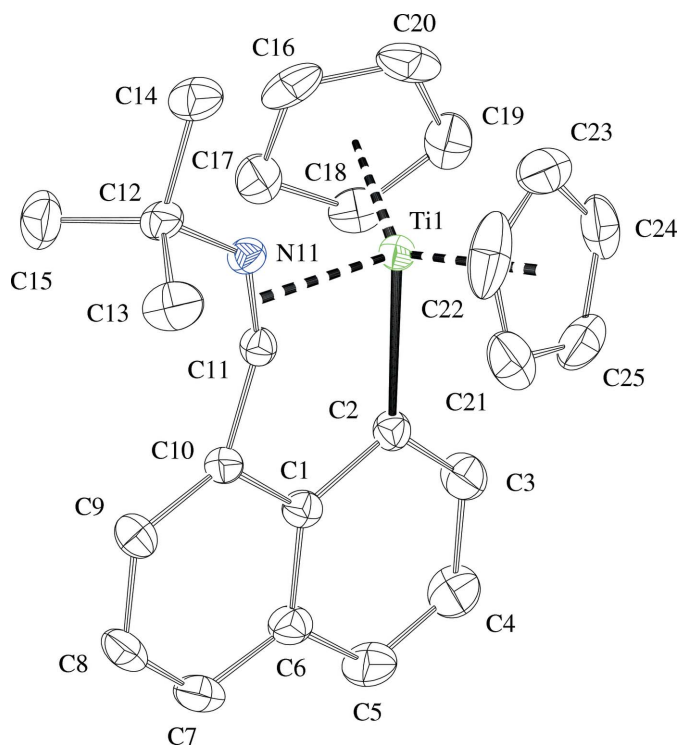
The structure of the title compound, (I), was determined in the course of our investigations on the synthesis and reactivity of small metallacycles. A similar compound has been reported by Campora *et al.* (1995) and a related compound with Zr by Berg & Petersen (1989).



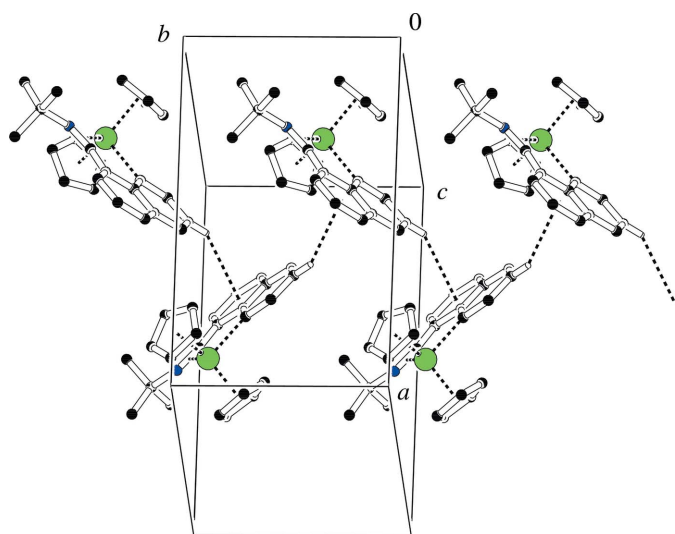
The structure of (I) is shown in Fig. 1, and selected bond lengths and angles are listed in Table 1. The title compound has two  $\eta^5$ -coordinated cyclopentadienyl (Cp) rings coordinated to the  $\text{Ti}^{\text{IV}}$  ion. The distances between the perpendicular projection of the  $\text{Ti}^{\text{IV}}$  ion on the Cp rings and their ring centroids (ring slippage) are 0.043 and 0.017  $\text{\AA}$  for the rings containing C16 and C21, respectively. The title compound further displays an iminoacyl fragment in an N-outside conformation, indicating that the  $\text{C}=\text{N}$  vector points away from the  $\sigma$ -bonded atom C2 (Tatsumi *et al.*, 1985). When the Cp rings and the  $\text{C}=\text{N}$  group are represented by their geometrical centres, as is suggested in Fig. 1, the  $\text{Ti}^{\text{IV}}$  ion displays a severely distorted tetrahedral configuration, with  $X-\text{Ti}-Y$  angles in the range  $88.24(11)$ – $133.16(9)^\circ$  (Table 3). In an alternative description, the  $\text{Ti}^{\text{IV}}$  ion takes part in two adjacent titanacycles, the three-membered ring  $\text{Ti1/C11/N11}$  and the five-membered ring  $\text{Ti1/C2/C1/C10/C11}$ . The maximum deviation from the least-squares plane through this last ring system is 0.028 (3)  $\text{\AA}$  for C2. The  $\text{Ti1}-\text{C2}$  bond is elongated by 0.2  $\text{\AA}$  compared with its parent compound 1,8-naphthalenediyltitanocene (Tinga *et al.*, 1994). The crystal packing displays a short  $\text{C}-\text{H}\cdots\pi$  contact (Fig. 2 and Table 2),

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**Figure 1**  
A view of the title compound, showing 30% probability displacement ellipsoids. H atoms have been omitted for clarity.



**Figure 2**  
Section of the crystal packing illustrating the C4—H4... $\pi$  interactions, viewed approximately perpendicular to the (102) planes. H atoms, with the exception of H4, have been omitted.

which links the molecules into an infinite chain running along the *b*-axis direction.

## Experimental

A solution of 1,8-naphthalenediyltitanocene (0.1 mmol), prepared according to Tinga *et al.* (1994), in pentane (10 ml) was added to a solution of CN-<sup>*t*</sup>Bu (0.2 mmol) in toluene (2 ml) at room temperature. After stirring for 3 h, the solvents were evaporated and the residue was extracted with cyclopentane. Brown crystals were obtained after

a week from a solution in cyclopentane, which was kept at 280 K. Spectroscopic analysis: <sup>1</sup>H NMR (200 MHz, room temperature, C<sub>6</sub>D<sub>6</sub>, reference C<sub>6</sub>D<sub>5</sub>H = 7.17 p.p.m., p.p.m.):  $\delta$  7.85 (*m*, 1 H, H3), 7.83 (*m*, 1 H, H9), 7.68 (*m*, 1 H, H5), 7.58 (*m*, 1 H, H4), 7.50 (*m*, 1 H, H7), 7.37 (*m*, 1 H, H8), 5.20 (*s*, 10 H, 2Cp), 1.21 (*s*, 9 H, <sup>*t*</sup>Bu) (at 240 K a similar proton NMR spectrum was measured in CD<sub>2</sub>Cl<sub>2</sub>); <sup>13</sup>C NMR (50.3 MHz, room temperature, C<sub>6</sub>D<sub>6</sub>, reference C<sub>6</sub>D<sub>6</sub> = 128.0 p.p.m., p.p.m.):  $\delta$  220.2 (*d*, <sup>3</sup>*J* = 3.3 Hz, C11), 191.2 (*m*, C2), 158.1 (*m*, C10), 140.3 (*ddd*, <sup>1</sup>*J* = 154.1 Hz, <sup>2</sup>*J* = 1.2 Hz, <sup>3</sup>*J* = 7.5 Hz, C3), 137.3 (*d*, <sup>3</sup>*J* = 7.6 Hz, C1 or C6), 135.0 (*dd*, <sup>3</sup>*J* = 8.0/6.6 Hz, C6 or C1), 132.0 (*d*, <sup>1</sup>*J* = 164.0 Hz, C9), 126.9 (*dd*, <sup>1</sup>*J* = 150.9 Hz, <sup>2</sup>*J* = 2.3 Hz, C4), 124.3 (*dd*, <sup>1</sup>*J* = 158.4 Hz, <sup>2</sup>*J* = 1.9 Hz, C8), 121.6 (*dd*, <sup>1</sup>*J* = 157.3 Hz, <sup>3</sup>*J* = 8.9 Hz, C7), 121.1 (*ddd*, <sup>1</sup>*J* = 157.3 Hz, <sup>3</sup>*J* = 9.4/8.2 Hz, C5), 105.5 (*dm*, <sup>1</sup>*J* = 171.0 Hz, 2Cp), 59.2 (*d*, <sup>2</sup>*J* = 4.2 Hz, C12), 29.3 (qsept, <sup>1</sup>*J* = 125.4 Hz, <sup>3</sup>*J* = 4.7 Hz, <sup>*t*</sup>Bu); IR (hexane)  $\bar{\nu}$  (cm<sup>-1</sup>) 1726 (*s*); HRMS (EI) calculated for C<sub>21</sub>H<sub>16</sub>N<sup>48</sup>Ti: 330.0764; found: 330.076; MS (EI) *m/z* (relative intensities): 387 (not observed, *M*<sup>+</sup>), 330 (77, [*M* - <sup>*t*</sup>Bu]<sup>+</sup>), 303 (17, [*M* - CHN-<sup>*t*</sup>Bu]<sup>+</sup>), 178 (38, Cp<sub>2</sub>Ti<sup>+</sup>), 128 (100, naphthalene, impurity).

## Crystal data

[Ti(C<sub>5</sub>H<sub>5</sub>)<sub>2</sub>(C<sub>15</sub>H<sub>15</sub>N)]  
*M<sub>r</sub>* = 387.33  
Monoclinic, *P*2<sub>1</sub>/*c*  
*a* = 15.0891 (12) Å  
*b* = 8.4475 (12) Å  
*c* = 15.6025 (18) Å  
 $\beta$  = 94.200 (13)°  
*V* = 1983.4 (4) Å<sup>3</sup>

*Z* = 4  
*D<sub>x</sub>* = 1.297 Mg m<sup>-3</sup>  
Mo *K*α radiation  
 $\mu$  = 0.44 mm<sup>-1</sup>  
*T* = 298 K  
Block, brown  
0.5 × 0.4 × 0.2 mm

## Data collection

Enraf-Nonius CAD-4  
diffractometer  
 $\omega/2\theta$  scans  
Absorption correction: none  
5661 measured reflections  
4533 independent reflections

2344 reflections with *I* > 2σ(*I*)  
*R*<sub>int</sub> = 0.200  
 $\theta_{\max}$  = 27.5°  
3 standard reflections  
frequency: 60 min  
intensity decay: 1.5%

## Refinement

Refinement on *F*<sup>2</sup>  
*R* [*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.056  
*wR* (*F*<sup>2</sup>) = 0.131  
*S* = 1.00  
4533 reflections  
247 parameters

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0508P)^2]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
( $\Delta/\sigma$ )<sub>max</sub> < 0.001  
 $\Delta\rho_{\max} = 0.28$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.28$  e Å<sup>-3</sup>

**Table 1**

Selected geometric parameters (Å, °).

Ti1—N11	2.197 (3)	Ti1—C11	2.043 (3)
Ti1—C2	2.292 (3)	N11—C11	1.238 (4)
N11—Ti1—C2	104.45 (11)	C11—N11—C12	132.5 (3)
N11—Ti1—C11	33.70 (12)	N11—C11—C10	150.8 (3)
C2—Ti1—C11	70.75 (12)		

**Table 2**

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C4—H4...Cg4 <sup>i</sup>	0.93	2.98	3.768 (5)	143

Symmetry code: (i)  $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$ . Cg4 is the centroid of the C1/C6/C7/C8/C9/C10 ring.

**Table 3**

Tetrahedral geometry of Ti1 (angles in °).

Cg1 represents the centroid of the Cp ring C16–C20; Cg2 represents the centroid of the Cp ring C21–C25; Cg3 represents the mid-point of the N=C bond.

C2–Ti1–Cg1	103.22 (10)
C2–Ti1–Cg2	101.46 (11)
C2–Ti1–Cg3	88.24 (11)
Cg1–Ti1–Cg2	133.16 (9)
Cg1–Ti1–Cg3	110.35 (9)
Cg2–Ti1–Cg3	109.60 (10)

H atoms were introduced in calculated positions, with C–H = 0.93 and 0.96 Å, and refined as riding on their carrier atoms, with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{methyl C})$  or  $1.2U_{\text{eq}}(\text{other C})$ . The methyl groups were refined as rigid groups, allowing for rotation around the C–C bond.

Data collection: locally modified *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *SET4* (de Boer & Duisenberg, 1984); data reduction: *HELENA* (Spek, 1997); program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985); program(s) used to refine struc-

ture: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *PLATON*.

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